**Objectives**

Upon completion of this activitiy, you will:

* Understand how chemical equilibrium explains how increasing CO2 in atmosphere and increasing global temperatures can affect the acidity of the oceans.

**Causal Principles**

1. Gravitational energy, thermal energy and/or chemical **energy** drive all movement and change of matter on Earth.

2. A system is in **equilibrium** when energy in the system is balanced.

3. Matter moves and changes to return a system to **equilibrium.**

4. **Energy** is needed to break bonds and is released when bonds form.

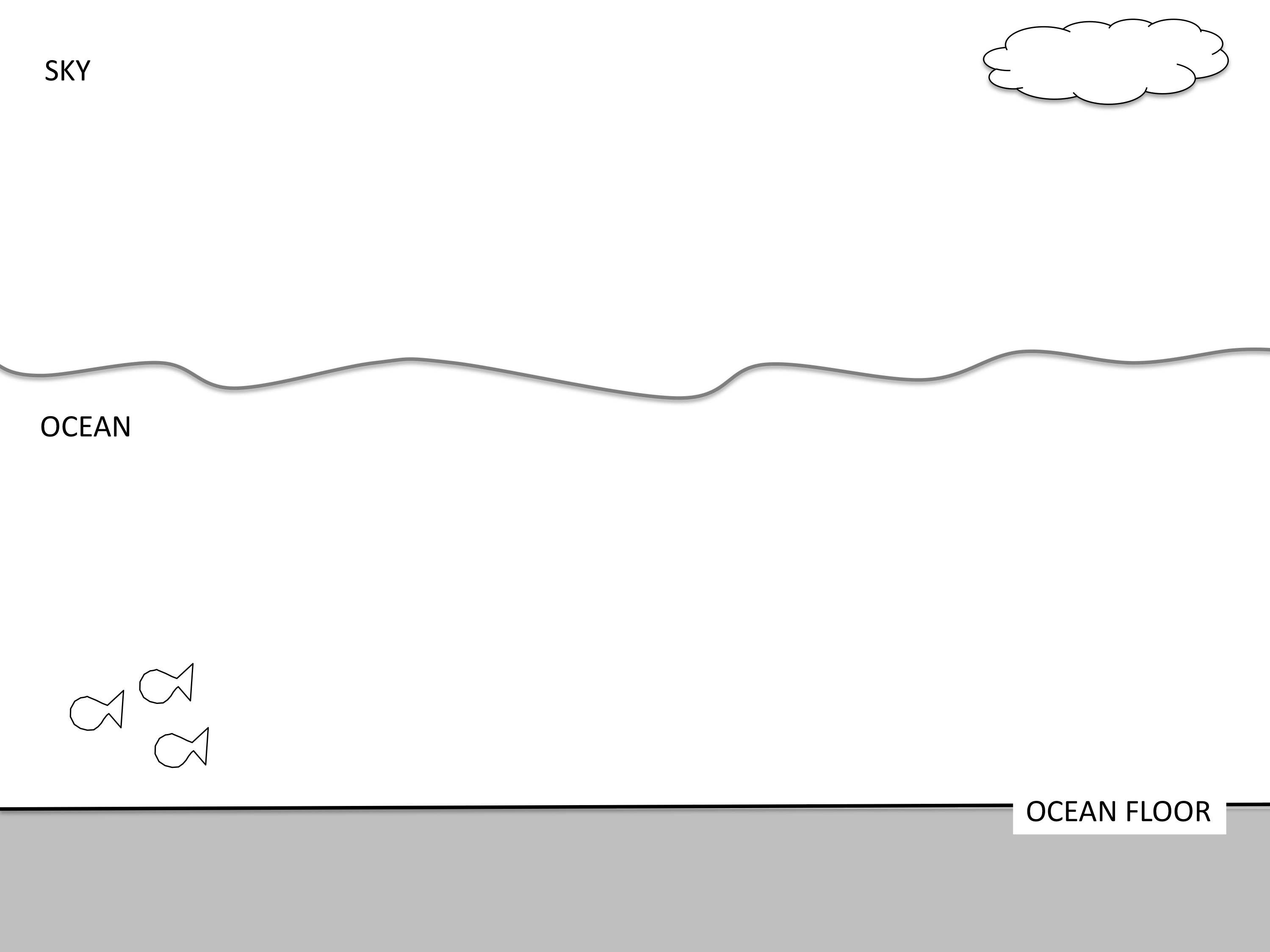
5. **Temperature** is a measure of the movement of molecules. Higher temperature means molecules are moving faster.

**PART 1: Background Notes**

**Class Notes**

On the diagram below, draw in the steps of the process of ocean acidification:

* Increased CO2 in the atmosphere
* Increased CO2 in the oceans
* Formation of hydrogen and bicarbonate ions



**Part 2: Group Work**

1. How does temperature of ocean water affect the acidity of the oceans?
2. Imagine the Sun begins to emit more light, resulting in a warming of Earth’s atmosphere. How would ocean acidification be affected?
3. Explain why increasing carbon dioxide in the atmosphere may lead to increased ocean acidification.

**Part 3: Homework**

If you complete the group work, you may work on the homework **on your own.** This means your answers should be generally unique from other students’ answers. **Submit your homework using ANGEL**.

1. You own a forest and decide to cut it down to build upscale condos. How could this effect the acidification of the oceans?
2. Extra nutrients are coming into the ocean due to waste from factory farming. The extra nutrients are causing algal blooms. Algae are known to use up carbon dioxide during growth. How would an increase in factory farming impact the acidity of the oceans?